

**PCT**WORLD INTELLECTUAL PROPERTY ORGANIZATION  
International Bureau

## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification <sup>6</sup> : <b>C09D 11/00</b>	A1	(11) International Publication Number: <b>WO 97/47697</b> (43) International Publication Date: 18 December 1997 (18.12.97)
(21) International Application Number: PCT/US97/08049	(81) Designated States: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, GH, HU, IL, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ, TM, TR, TT, UA, UG, UZ, VN, YU, ARIPO patent (GH, KE, LS, MW, SD, SZ, UG), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG).	
(22) International Filing Date: 13 May 1997 (13.05.97)		
(30) Priority Data: 08/665,127 14 June 1996 (14.06.96) US		
(71) Applicant: CABOT CORPORATION [US/US]; 75 State Street, Boston, MA 02109-1806 (US).		
(72) Inventors: BATES, Jodi, A.; 19 Chatham Street, Billerica, MA 01821 (US). JOHNSON, Joseph, E.; 9 Ayer Street, Nashua, NH 03060 (US).	Published <i>With international search report.</i>	
(74) Agent: LANDO, Michelle, B.; Cabot Corporation, 157 Concord Road, P.O. Box 7001, Billerica, MA 01821-7001 (US).		

**(54) Title:** INK COMPOSITIONS HAVING IMPROVED LATENCY**(57) Abstract**

An ink composition is disclosed including a liquid vehicle and an ionically charged organic pigment and a hydroxylated hydrocarbon humectant having at least two hydroxyl groups. The ionically charged pigment may be black, blue, brown, cyan, green, violet, magenta, red, orange, yellow, mixtures thereof, and the like. The ionically charged pigment, due to the nature of its surface charge, is readily dispersed in the liquid vehicle of the ink composition without the addition of a surfactant or other dispersing aid or additive. In a preferred embodiment, the ionically charged pigment has sufficient ionizable groups to maintain the colloidal stability of the pigment in the ink composition. Further disclosed is a method for generating printed images which include incorporating into a printing apparatus the ink compositions described above and generating an image onto a substrate.

***FOR THE PURPOSES OF INFORMATION ONLY***

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav Republic of Macedonia	TM	Turkmenistan
BF	Burkina Faso	GR	Greece			TR	Turkey
BG	Bulgaria	HU	Hungary	ML	Mali	TT	Trinidad and Tobago
BJ	Benin	IE	Ireland	MN	Mongolia	UA	Ukraine
BR	Brazil	IL	Israel	MR	Mauritania	UG	Uganda
BY	Belarus	IS	Iceland	MW	Malawi	US	United States of America
CA	Canada	IT	Italy	MX	Mexico	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NE	Niger	VN	Viet Nam
CG	Congo	KE	Kenya	NL	Netherlands	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NO	Norway	ZW	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's Republic of Korea	NZ	New Zealand		
CM	Cameroon	KR	Republic of Korea	PL	Poland		
CN	China	KZ	Kazakhstan	PT	Portugal		
CU	Cuba	LC	Saint Lucia	RO	Romania		
CZ	Czech Republic	LI	Liechtenstein	RU	Russian Federation		
DE	Germany	LK	Sri Lanka	SD	Sudan		
DK	Denmark	LR	Liberia	SE	Sweden		
EE	Estonia			SG	Singapore		

TITLE**INK COMPOSITIONS HAVING IMPROVED LATENCY**BACKGROUND OF THE INVENTION1. Field of the Invention.

5       The present invention is directed to ink compositions and, more particularly, to aqueous ink compositions suitable for imaging applications, such as ink jet printing processes.

2. Discussion of the Related Art.

10      Ink jet printing is a non-impact process wherein droplets of ink are produced and deposited on a substrate such as paper, transparent film, or textile material in response to an electronic signal. Ink jet printing systems are typically classified by two known types: continuous stream or drop-on-demand.

15      Ink compositions which are useful in imaging applications, such as ink jet ink printing systems, are well known and generally contain water soluble dyes. Also known is the addition of suitable cosolvents systems such as glycols and other humectants to such dye based ink compositions in order to improve the latency and recoverability in printers utilizing such compositions. See, for example, U.S. Patent No. 5,389,133 to Gundlach et al., U.S. Patent No. 5,286,289 to Winnik et al., U.S. Patent No. 5,271,764 to Winnik et al., U.S. Patent No. 4,840,674 to Schwartz, the specifications of which are incorporated herein in its entirety by reference.

20      Although dye-based inks are suitable for their intended purposes, dyes have several disadvantages when used in ink jet inks. For example, dyes, being water-

- 2 -

soluble in a water/organic mixture, may dissolve and run when exposed to moisture or water. Dye images may further smear or rub off on contact with felt pen markers or upon being rubbed or touched by finger. Dyes also exhibit poor light stability when exposed to visible or ultraviolet light.

5           Pigments are also known as colorants in ink compositions but have not received a wide degree of acceptance in ink jet ink systems, for example, because of problems associated with the performance and reliability of the composition, i.e., print properties, stability, latency, and the like.

10         As a result, although known compositions are suitable for their intended purpose, a need remains for improved ink compositions, especially for use in the ink jet printers, which overcome the problems typically associated with current dye-based and pigment system. In addition, there is a need for improved ink compositions which exhibit improved latency and recoverability in their respective printing systems while providing good print properties.

15          SUMMARY OF THE INVENTION

20         The present invention is directed to ink compositions comprising an ionically charged pigment and a hydroxylated hydrocarbon humectant having at least two hydroxyl groups. The ionically charged pigment includes, but is not limited to, black, blue, brown, cyan, green, violet, magenta, red, orange, yellow, as well as mixtures thereof. The hydroxylated hydrocarbon humectant is selected from the group consisting of glycols, glycerol, polyols, and derivatives and mixtures thereof.

Also disclosed is a method for generating printed images which include incorporating into a printing apparatus the ink compositions described above and generating an image onto a substrate.

DETAILED DESCRIPTION OF THE INVENTION

The present invention is directed to ink compositions comprising an ionically charged pigment and a hydroxylated hydrocarbon humectant having at least two hydroxyl groups.

5       The ionically charged pigment includes, but is not limited to, black, blue, brown, cyan, green, violet, magenta, red, orange, yellow, mixtures thereof, and the like. By ionically charged is meant that the pigment, due to the nature of its surface groups, is readily-dispersing in the liquid vehicle of the ink composition without the addition of a surfactant or other dispersing aid or additive. Preferably, the ionically charged pigment further has sufficient ionizable groups which will enable the ink compositions to further maintain its colloidal stability without the addition of a surfactant or stabilizing aid. As a result, the ionically charged pigment of the present invention will readily disperse, and remains dispersed, in the liquid vehicle of the ink composition under low shear stirring or mixing conditions.

15      Examples of suitable black pigments include carbon blacks, such as channel black, furnace black, lamp black and the like. Examples of suitable classes of colored pigments include anthraquinones, monoazos, disazos, phthalocyanine blues, phthalocyanine greens, pyranthrone, perylenes, heterocyclic yellows, quinacridones and indigoids.

20      In a preferred embodiment, the ionically charged pigment has an attached organic group. A preferred set of organic groups which may be attached are organic groups substituted with an ionic or an ionizable group as a functional group. An ionizable group is one capable of forming an ionic group in the medium of use. The ionic group may be an anionic group or a cationic group and the ionizable group may form an anion or cation.

Ionizable functional groups forming anions include, for example, acidic groups or salts of acidic groups. The organic groups, therefore, include groups derived from organic acids. Preferably, when an organic group contains an ionizable group forming an anion, such an organic group has a) an aromatic group and b) at least one acidic group having a  $pK_a$  of less than 11, or at least one salt of an acidic group having a  $pK_a$  of less than 11, or a mixture of at least one acidic group having a  $pK_a$  of less than 11 and at least one salt of an acidic group having a  $pK_a$  of less than 11. The  $pK_a$  of the acidic group refers to the  $pK_a$  of the organic group as a whole, not just the acidic substituent. More preferably, the  $pK_a$  is less than 10 and most preferably less than 9. Preferably, the aromatic group of the organic group is directly attached to the carbon black. The aromatic group may be further substituted or unsubstituted, for example, with alkyl groups. More preferably, the organic group is a phenyl or a naphthyl group and the acidic group is a sulfonic acid group, a sulfinic acid group, a phosphonic acid group, or a carboxylic acid group. Most preferably, the organic group is a substituted or unsubstituted sulfophenyl group or a salt thereof; a substituted or unsubstituted carboxyphenyl; a substituted or unsubstituted (polysulfo)phenyl group or a salt thereof; a substituted or unsubstituted sulfonaphthyl group or a salt thereof; or a substituted or unsubstituted (polysulfo)naphthyl group or a salt thereof. Such groups include, for example, C<sub>6</sub>H<sub>4</sub>SO<sub>3</sub><sup>-</sup>M<sup>+</sup>, C<sub>6</sub>H<sub>4</sub>CO<sub>2</sub><sup>-</sup>M<sup>+</sup>, wherein M<sup>+</sup> is Na<sup>+</sup>, K<sup>+</sup>, or Li<sup>+</sup>.

Amines represent examples of ionizable functional groups that form cationic groups. For example, amines may be protonated to form ammonium groups in acidic media. Preferably, an organic group having an amine substituent has a  $pK_b$  of less than 5. Quaternary ammonium groups and quaternary phosphonium groups also represent examples of cationic groups. Preferably, the organic group contains an aromatic group such as a phenyl or a naphthyl group and a quaternary ammonium or a quaternary phosphonium group. The aromatic group is preferably directly attached to the carbon black. Quaternized cyclic amines, and even quaternized aromatic

- 5 -

amines, can also be used as the organic group. Thus, N-substituted pyridinium compounds, such as N-methyl-pyridyl, can be used in this regard.

Examples of organic groups include, but are not limited to,  $C_6H_4N(CH_3)_3^+X^-$ ,  $C_6H_4COCH_2N(CH_3)_3^+X^-$ ,  $C_6H_4(NC_5H_5)^+X^-$ ,  $(C_5H_4N)C_2H_5^+X^-$ , 5  $C_6H_4COCH_2(NC_5H_5)^+X^-$ ,  $(C_5H_4N)CH_3^+X^-$ , and  $C_6H_4CH_2N(CH_3)_3^+X^-$ , wherein  $X^-$  is a monovalent ion.

The ionically charged pigment of the present invention is prepared by the method described in U.S. Patent Application Serial No. 08/356,653 to Belmont entitled, "Reaction of Carbon Materials with Diazonium Salts and Resultant Carbon Products," filed December 15, 1994; U.S. Patent Application No. 08/356,660, to 10 Belmont entitled, "Reaction of Carbon Black with Diazonium Salts, Resultant Carbon Black Products and Their Uses," filed December 15, 1994; and U.S. Patent Application No. 08/527,525 to Belmont entitled, "Reaction of Carbon Black with Diazonium Salts, Resultant Carbon Black Products and Their Uses," filed December 15, 1995, the disclosures of which are fully incorporated herein by reference. Although such methods are primarily directed to carbon materials and carbon black, 15 such described methods are useful to prepare the ionically charged pigment of the present invention, whether it be a black or colored pigment. It is also recognized by those skilled in the art that other methods of preparation which would yield the required ionic stabilization are also suitable.

The ionically charged pigment of the present invention is present in the ink compositions in an amount effective to provide the desired image quality, e.g., optical density, without detrimentally affecting the performance of the ink. Typically, the ionically charged pigment will be present in an amount ranging from about 1% to 25 about 20%, preferably from about 2% to about 10%, based on the weight of the ink composition. In addition, if the ionically charged pigment is a modified pigment having at least one attached organic group, the pigment should be treated with the

desired modifying agent in an amount sufficient to provide the desired utility in the end product application while maintaining the colloidal stability of the pigment in the ink composition.

5        The ionically charged pigment is typically as small as possible to enable a stable colloidal suspension of the pigment in the liquid vehicle and to prevent clogging of the ink channels and nozzles when used in the desired printing application. For example, a preferred average aggregate diameter of the ionically charged pigment for use in a thermal ink jet printer are generally below 1.0 micron, preferably in a range from about 0.005 micron to about 0.3 micron.

10      The humectant of the present invention is characterized as a hydroxylated hydrocarbon having at least two hydroxyl groups. The humectant should have a sufficient degree of miscibility and solubility in the liquid vehicle of the ink composition to provide improved latency of the ink composition while maintaining the colloidal stability of the system. Examples of suitable humectants include glycols and derivatives thereof, such as ethylene glycol, butylene glycol, propylene glycol, diethylene glycol, dipropylene glycol, pentamethylene glycol, polyethylene glycol, polypropylene glycol, trimethylene glycol; glycerol and derivatives thereof; polyols and derivatives thereof, such as butanetriol, trishydroxymethylethane, mesoerythritol, and xylitol; as well as mixtures thereof.

15

20      The humectant of the present invention is present in an amount effective to improve the latency of the ink composition in a printing apparatus without detrimentally effecting the other properties of the ink composition, e.g. viscosity, conductivity, drop size, pH, optical density, surface tension and the like. Generally, depending on the particular humectant used, the amount of the humectant in the ink compositions is equal to or greater than about 15 %, by weight, and preferably equal to or greater than 20 %, by weight, because such systems were found to possess good latency, stability and physical properties.

25

Although the use of humectants are known in the prior art, it has been found that many such humectants, e.g. alcohols and ethers, are not suitable with the ionically charged pigment of the present invention due to insufficient latency and colloidal instability. In addition, the humectants of the prior art are effectively used at levels considerably lower than that disclosed herein. It has therefore been recognized that ionically charged pigment systems will behave differently than existing dyes and nonionically charged pigment (i.e. pigment systems requiring the addition of a surfactant or polymer for dispersability). In order to achieve a useful ink system with desired properties, a unique interrelationship has been found to exist between the ionically charged pigment itself, the type of humectant, and the amount of humectant in a particular ink composition.

The ink compositions of the present invention experience improved latency and recoverability when used in printing systems. Latency is the period during which the printer may stop printing while it contains the ink and subsequently be restarted without clogging of the printer nozzle. Stated differently, latency is the period of time during which the printhead can remain idle before deterioration of the inks' jetting performance occurs. As a result, the latency should be as high, i.e. long, as possible to enable restarting of the printer after extended idle periods.

The ink compositions of the present inventions may be prepared utilizing conventional techniques known to those skilled in the art, such as combining or mixing the desired component in a suitable liquid vehicle or medium. Typically, the ink compositions are aqueous systems and, as such, include a significant amount of water, preferably deionized or distilled water, or water and a water miscible or water soluble solvent. For example, the amount of water or similar medium is generally present in an amount ranging from about 55% to about 80%, preferably from about 65% to 75%, based on the weight of the ink composition.

The ink compositions of the present invention may be buffered to the desired pH by the addition of a suitable base, such as sodium hydroxide, ammonium hydroxide, triethylamine, dimethylethanolamine and the like, or a suitable acid, such as mineral acids, hydrochloric acid, sulfuric acid and the like. In addition, a polymer which is soluble in the ink composition may be added to improve the waterfastness of the images generated from the ink compositions. By "soluble" is meant that the polymer will dissolve in the ink vehicle to form a one phase system. These polymers include, for example, polyvinyl alcohol, polyester, polyestermelamine, styrene-acrylic acid copolymers, styrene-maleic acid copolymers, styrene-maleic acid-alkyl acrylate copolymers, styrene-metacrylic acid copolymers, styrene-metacrylic acid-alkyl acrylate copolymers, styrene-maleic half ester copolymers, vinyl napthalene-acrylic acid copolymers, vinyl napthalene-maleic acid copolymers and salts thereof. Additional polymers include polyvinylimidazole, derivatives of polyvinylimidazole, copolymers of vinylimidazole, copolymers of vinylimidazole derivatives, polyvinylpyridine, derivatives of polyvinylpyridine, copolymers of vinylpyridine, copolymers of vinylpyridine derivatives, polyethyleneimine, derivatives of polyethyleneime, and mixtures thereof, as described in U. S. Patent Application Serial No. 08/628,202 to Bates et al., entitled "Ink Compositions and Method of Generating Images Produced Therefrom," the disclosure of which is fully incorporated herein by reference.

Suitable additives are also generally incorporated into the ink compositions to impart a number of desired properties while maintaining the stability of the compositions. For example, a surfactant or suitable polymer may be used to enhance the colloidal stability of the ionically charged pigment in the ink composition. Other additives are well known in the art and include biocides, binders, drying accelerators, penetrants and the like. The amount of a particular additive will vary depending on a variety of factors but are generally present in an amount ranging between 0% and 10%, preferably between 0.1% and 5%, although the amount may be outside this range.

Printed images may be generated from the ink compositions of the present invention by incorporating such compositions into a suitable printing apparatus, and generating an image onto a substrate. Suitable ink jet printers include, for example, thermal printers, piezoelectric printers, continuous printers, valve printers and the like. Similarly, any suitable substrate can be employed including plain papers, bonded papers, coated papers, transparency materials, textile materials, plastics, polymeric films, inorganic substrates and the like.

Non-limiting illustrations of the present invention follow.

#### EXAMPLE 1

A modified carbon black having  $C_6H_4SO_3^-Na^+$  surface groups was prepared by charging a continuous pin mixer with 100 parts per hour of a carbon black having a BET surface area of 200  $m^2/g$  (BET surface area determined according to ASTM D-4820) and DBPA of 122 mL/100 g (DBPA structure data determined according to ASTM D-2414), 12.5 parts per hour of sulfanilic acid and 5.2 parts per hour of sodium nitrite as an aqueous solution. The resultant material was a mixture of an ionically charged modified carbon black and water containing 60% solids. The modified carbon black product was diluted with water and filtered before use.

Ink compositions were prepared by mixing a 20%, by weight, dispersion of the ionically charged modified carbon black having  $C_6H_4SO_3^-Na^+$  surface groups with varying concentrations of glycerol (available from Aldrich Chemical Co., Milwaukee, WI) and the remainder distilled water. The resultant inks contained 5%, by weight, of the ionically charged modified carbon black. Other relevant ink properties of the resulting inks are illustrated in Table I.

- 10 -

Table I

Sample	% glycerol in ink	Viscosity* (cps)	Surface Tension† (dynes/cm)	pH
5	1	1.7	74.5	7.8
	2	2.0	74.0	7.1
	3	2.4	74.0	7.0
	4	3.2	73.0	7.0
	5	4.2	72.0	7.2
	6	6.2	72.0	7.1

10 \*Viscosity was measured using a No. 1 and No. 2 shell cup from Norcross Corporation, Newton, Massachusetts

15 †Surface tension was measured using a CSC DU NOUY Tensiometer, No. 70535 from CSC Scientific Co., Inc. Fairfax, Virginia using ASTM D-1331 and D-1590 procedures.

The ink compositions were placed in Hewlett Packard 51626A cartridges  
 15 (which had been previously emptied and cleaned) and tested for latency using a Hewlett-Packard DeskJet® 540 ink jet printer. The cartridges were removed from the printer and placed uncapped in an upside down position for a time period of 24 hours, 72 hours and 128 hours. After each time period, the cartridges were placed back in the printer and test patterns were generated. Latency is a measure of the amount of time for which a flow of ink through a nozzle can be stopped and later re-started without clogging. In this example, the latency of the inks was measured as a function of: 1) the percentage of firing nozzles after printing one test pattern; and 2) the number of pages required to yield a 100% firing. After the initial test pattern, a printer nozzle check, as described in the manufacturer's literature, was printed to determine the percentage of firing nozzles. Five additional test patterns were then  
 20  
 25

- 11 -

printed and the point at which all nozzles began to fire was recorded. Excellent latency values would be 100% of the nozzles firing immediately, i.e. recoverability=0 pages. Latency results of the ink compositions are illustrated in Table II.

5

Table II

Sample	24 Hour Latency		72 Hour Latency		168 Hour Latency	
	% firing nozzles	Recoverability** (pages)	% firing nozzles	Recoverability** (pages)	% firing nozzles	Recoverability** (pages)
1	0	None	no further testing			
2	0	None	no further testing			
3	50	None	no further testing			
10	100	0	100	0	100	0
4	100	0	100	0	100	0
5	100	0	100	0	100	0
6	100	0	100	0	100	0

\*\*Recoverability = pages required to yield 100% nozzle firing.

None = no recovery. i.e. the nozzle remained clogged after 6 pages were printed.

15

Partial Recovery = some clogged nozzles recovered, but not all.

0 pages - all nozzles fired immediately.

20

As shown in Table II, the ink compositions of samples 3 through 6 containing greater than 20% humectant and, in particular, equal or greater than 30% humectant demonstrated enhanced latency. Samples 4 through 6 showed 100% nozzle firing on 0 pages.

## EXAMPLE 2

Ink compositions were prepared by repeating the procedure of Example 1, except that varying concentrations of ethylene glycol (available from Aldrich Chemical Co., Milwaukee, WI). The resultant inks similarly contained 5%, by

- 12 -

weight, of the modified carbon black. Other relevant ink properties of the resulting inks are illustrated in Table III.

Table III

Sample	% ethylene glycol in ink	Viscosity (cps)	Surface Tension (dynes/cm)	pH
1	0	1.7	74.5	7.8
2	10	2.2	72.0	7.3
3	20	2.5	68.0	7.3
4	30	2.9	64.5	7.2

The ink compositions were placed in Hewlett Packard 51626A cartridges and tested for latency utilizing the procedure described in Example 1. Latency results of the ink compositions are illustrated in Table IV.

Table IV

Sample	24 Hour Latency		72 Hour Latency		168 Hour Latency	
	% firing nozzles	Recoverability** (pages)	% firing nozzles	Recoverability* (pages)	% firing nozzles	Recoverability* (pages)
1	0	None	no further testing			
2	0	None	no further testing			
3	81	1.25	98	1.25	4	None
4	100	0	100	0	100	0

\*\*Recoverability = pages required to yield 100% nozzle firing.

None = no recovery. i.e. the nozzle remained clogged after 6 pages were printed.

Partial Recovery = some clogged nozzles recovered, but not all.

0 pages - all nozzles fired immediately.

- 13 -

As illustrated in Table IV, the ink compositions of samples 3 and 4 containing equal or greater than 20% of ethylene glycol demonstrated an improvement in latency when compared to ink compositions containing less than 20% ethylene glycol or no humectant.

5

### EXAMPLE 3

Ink compositions were prepared by repeating the procedure of Example 1, except that various humectants (available from Aldrich Chemical Co., Milwaukee, WI) in accordance with the present invention was substituted for the glycerol. The resultant inks similarly contained 5%, by weight, of the modified carbon black.  
 10 Other relevant ink properties of the resulting inks are illustrated in Table V.

Table V

Sample	% humectant	humectant	Viscosity (cps)	Surface Tension (dynes/cm)	pH
1	0	NA	1.7	74.5	7.8
2	30	1,3 propanediol	3.9	54.0	7.0
15	3	25 1,4 butanediol	3.7	58.5	6.9
4	20	1,5 petanediol	3.0	54.0	6.9
5	30	1,2,4 butanetriol	3.6	67.0	6.0
6	30	1,1,1tris(hydroxy-methyl) ethane	4.1	65.0	7.5
7	20	mesoerythritol	2.4	75.5	7.2
20	8	xylitol	3.2	77.0	7.1

- 14 -

5           The concentration of the humectant in the ink compositions of Table V were chosen based on the compatibility with the pigment dispersion of the modified carbon black. Compatibility was determined by mixing the pigment dispersion in varying concentrations of humectant. The resulting dispersion was evaluated using a light microscope at 400X magnification and compared to a neat pigment dispersion (i.e. no humectant). The ink compositions of samples 2-8 in Table V all exhibited a high level of compatibility.

10           The ink compositions were placed in Hewlett Packard 51626A cartridges and tested for latency after a 72 hour period utilizing the procedure described in Example 1. Latency results of the ink compositions are illustrated in Table VI.

Table VI

Sample	72 Hour Latency	
	% firing nozzles	Recoverability** (pages)
1	0	None
2	100	0
3	100	0
4	100	0
5	100	0
15           6	98	5
20           7	31	5
25           8	98	None

\*\*Recoverability = pages required to yield 100% nozzle firing.

None = no recovery. i.e. the nozzle remained clogged after 6 pages were printed.

Partial Recovery = some clogged nozzles recovered, but not all.

0 pages - all nozzles fired immediately.

- 15 -

All nozzles of the printhead containing the ink compositions of samples 2 through 5 all fired immediately and required zero (0) pages to recover. Although all the nozzles containing the ink composition of samples 6 and 7 did not fire immediately on page one, all nozzles recovered fully after five pages. Although one of the nozzles of the printhead did not recover after 6 pages, most of the nozzles containing the ink composition of sample 8 fired immediately. Table IV therefore illustrates that the ink compositions containing the varying humectants at concentrations equal or greater than 20% demonstrated an improvement in latency when compared to ink compositions lacking the humectant.

10

#### EXAMPLE 4

Ink compositions were prepared by repeating the procedure of Example 1, except that a mixture of ethylene glycol and glycerol (both available from Aldrich Chemical Co., Milwaukee, WI) was substituted for glycerol. The resultant inks similarly contained 5 %, by weight, of the modified carbon black. Other relevant ink properties of the resulting inks are illustrated in Table VII.

Table VII

Sample	% humectant	Viscosity (cps)	Surface Tension (dynes/cm)	pH
1	0	1.7	74.5	7.8
2	15% ethylene glycol & 15% glycerol mixture	3.0	68.0	7.2

- 16 -

The ink compositions were placed in Hewlett Packard 51626A cartridges and tested for latency after a 72 hour period utilizing the procedure described in Example 1. Latency results of the ink compositions are illustrated in Table VIII.

Table VIII

5

Sample	72 Hour Latency	
	% firing nozzles	Recoverability** (pages)
1	0	None
2	100	0

\*\*Recoverability = pages required to yield 100% nozzle firing.

10 None = no recovery. i.e. the nozzle remained clogged after 6 pages were printed.  
Partial Recovery = some clogged nozzles recovered, but not all.  
0 pages - all nozzles fired immediately.

15 As illustrated in Table VIII, the ink composition containing the ethylene glycol / glycerol mixture demonstrated an improvement in latency when compared to ink compositions lacking the humectant. As a result, it is expected that various mixtures of the humectant of the present invention would also be suitable as described herein.

#### EXAMPLE 5

20 Ink compositions were prepared by repeating the procedure of Example 1, except that varying levels of the modified carbon black was used with 30% glycerol (available from Aldrich Chemical Co., Milwaukee, WI). The relevant ink properties of the resulting inks are illustrated in Table IX.

- 17 -

Table IX

Sample	% carbon black	Viscosity (cps)	Surface Tension (dynes/cm)	pH
1	1	2.7	70.0	7.2
2	5	3.2	73.0	7.0
3	7	3.3	70.0	7.3
4	10	4.5	70.0	7.3

The ink compositions were placed in Hewlett Packard 51626A cartridges and tested for latency after a 72 hour period utilizing the procedure described in Example 1. Latency results of the ink compositions are illustrated in Table X.

10

Table X

Sample	72 Hour Latency	
	% firing nozzles	Recoverability** (pages)
1	100	0
2	100	0
3	100	0
4	100	0

\*\*Recoverability = pages required to yield 100% nozzle firing.

None = no recovery. i.e. the nozzle remained clogged after 6 pages were printed.

Partial Recovery = some clogged nozzles recovered, but not all.

0 pages - all nozzles fired immediately.

15  
20

- 18 -

As illustrated in Table X, the ink composition of the present invention may be used with various levels of the ionically charged pigment while maintaining an improvement in latency.

EXAMPLE 6

5 Ink compositions were prepared by repeating the procedure of Example 1, except that various bases were also added to the compositions to adjust the pH of the ink compositions to approximately 8.5. The relevant ink properties of the resulting inks are illustrated in Table XI.

Table XI

10	Sample	type of base	% glycerol	Viscosity (cps)	Surface Tension (dynes/cm)	pH
1	sodium hydroxide	30	3.0	70.0	8.5	
2	ammonium hydroxide	30	3.2	73.0	8.5	
3	triethylamine	30	3.2	70.0	8.5	
4	dimethylethanol-amine	30	3.2	70.0	8.7	

15 The ink compositions were placed in Hewlett Packard 51626A cartridges and tested for latency after a 72 hour period utilizing the procedure described in Example 1. Latency results of the ink compositions are illustrated in Table XII.

- 19 -

Table XII

Sample	72 Hour Latency	
	% firing nozzles	Recoverability** (pages)
5	1	0
	2	0
	3	0
	4	0

\*\*Recoverability = pages required to yield 100% nozzle firing.

None = no recovery. i.e. the nozzles remained clogged after 6 pages were printed.

Partial Recovery = some clogged nozzles recovered, but not all.

0 pages - all nozzles fired immediately.

The ink composition of Table XII all exhibited improved latency in accordance with the present invention.

15

#### EXAMPLE 7

A modified carbon black having  $C_6H_4NC_5H_5^+Cl^-$  surface groups was prepared by adding 8.31 g of silver nitrate to a solution of 11.2 g of N-(4-aminophenyl)pyridinium chloride in water. The mixture was stirred at 70°C-80°C for one hour and stored at room temperature overnight. The mixture was then heated at 70°C-80°C for an additional hour, cooled and filtered through a bed of diatomaceous earth. A portion of the resulting solution (53.5g) was added to 10 g of a carbon black having a CTAB surface area of 108 m<sup>2</sup>/g (CTAB surface area determined according to ASTM D-3765) and DBPA of 116 mL/100 g (DBPA structure data determined according to ASTM D-2414). Approximately 35 g of water and 0.7 g of concentrated HCl was added to the dispersion. The resulting dispersion

- 20 -

was stirred at 90°C for 2 hours and dried in an oven at 125°C to form an ionically charged modified carbon black having  $C_6H_4NC_5H_5^+Cl^-$  surface groups.

Ink compositions were prepared by mixing a 20%, by weight, dispersion of the ionically charged modified carbon black having  $C_6H_4NC_5H_5^+Cl^-$  surface groups with varying concentrations of glycerol (available from Aldrich Chemical Co., Milwaukee, WI) and the remainder distilled water. The resultant inks contained 5%, by weight, of the modified carbon black. Other relevant ink properties of the resulting inks are illustrated in Table XIII.

Table XIII

Sample	% glycerol in ink	Viscosity (cps)	Surface Tension (dynes/cm)	pH
1	10	2.1	74.0	3.0
2	20	2.6	74.0	3.0
3	30	3.2	73.0	2.9

The ink compositions were placed in Hewlett Packard 51626A cartridges and tested for latency utilizing the procedure described in Example 1. Latency results of the ink compositions are illustrated in Table XIV.

- 21 -

Table XIV

Sample	24 Hour Latency		72 Hour Latency		168 Hour Latency	
	% firing nozzles	Recoverability** (pages)	% firing nozzles	Recoverability** (pages)	% firing nozzles	Recoverability** (pages)
1	50	1.25	88	1.5	71	Partial Recovery
2	100	0.5	94	1.25	19	4
5	94	1.25	56	2.0	4	3

\*\*Recoverability = pages required to yield 100% nozzle firing.

None = no recovery, i.e. the nozzles remained clogged after 6 pages were printed.

Partial Recovery = some clogged nozzles recovered, but not all.

0 pages - all nozzles fired immediately.

10 Table XIV illustrates that ink compositions containing equal or greater than 20% glycerol demonstrated an improvement in latency when compared to ink compositions containing a lesser amount of humectant. For example, after a 168 hour latency period, all nozzles of the printhead recovered after 4 pages.

#### EXAMPLE 8

15 Ink compositions were prepared by repeating the procedure of Example 7, except that varying concentrations of ethylene glycol (available from Aldrich Chemical Co., Milwaukee, WI) was used as the humectant. The resultant inks similarly contained 5%, by weight, of the modified carbon black. Other relevant ink properties of the resulting inks are illustrated in Table XV.

- 22 -

Table XV

Sample	% ethylene glycol in ink	Viscosity (cps)	Surface Tension (dynes/cm)	pH
1	10	2.0	71.0	3.1
2	20	2.5	67.0	3.0
3	30	3.0	65.0	3.9

The ink compositions were placed in Hewlett Packard 51626A cartridges and tested for latency utilizing the procedure described in Example 1. Latency results of the ink compositions are illustrated in Table XVI.

Table XVI

Sample	24 Hour Latency		72 Hour Latency		168 Hour Latency	
	% firing nozzles	Recoverability** (pages)	% firing nozzles	Recoverability** (pages)	% firing nozzles	Recoverability** (pages)
1	75	Partial Recovery	0	None	Not further tested	
2	90	2	92	3	0	None
3	80	3	60	Partial Recovery	12	None

\*\*Recoverability = pages required to yield 100% nozzle firing.

None = no recovery. i.e. the nozzles remained clogged after 6 pages were printed.

Partial Recovery = some clogged nozzles recovered, but not all.

0 pages - all nozzles fired immediately.

As illustrated in Table XIV, ink compositions containing equal or greater than 20% of ethylene glycol demonstrated an improvement in latency when compared to ink compositions containing less than 20%, especially after 74 hours. It is recognized

that although the ethylene glycol does show improvement in latency, glycerol may be a preferred humectant, as shown in Example 7, with an ionically charged modified carbon black having  $C_6H_4NC_5H_5^+Cl^-$  surface groups.

5 As described herein, the ink compositions of the present invention demonstrated characteristics which make them desirable for many printing systems, especially for ink jet ink systems. Particularly, the ink compositions have improved latency and are expected to exhibit, when formulated in actual ink systems, good physical properties such as, pH, viscosity, and surface tension. In addition, the ink  
10 compositions of the present invention are further expected to produce quality images having good waterfastness.

15 It is further understood that the present invention is not limited to the particular embodiments shown and described herein, but that various changes and modifications may be made without departing from the scope and spirit of the invention.

What is claimed is:

CLAIMS

1. An ink composition, comprising a liquid vehicle and an ionically charged pigment and a hydroxylated hydrocarbon humectant having at least two hydroxyl groups, wherein said humectant is present in an amount sufficient to improve the latency of said ink composition in a printing apparatus.  
5
2. The ink composition of claim 1, wherein said ink composition is an aqueous ink composition.
3. The ink composition of claim 1, wherein said ionically charged pigment is selected from the group consisting of: black, blue, brown, cyan, green, violet, magenta, red, orange, yellow and mixtures thereof.  
10
4. The ink composition of claim 1, wherein said ionically charged pigment has at least one attached organic group.
5. The ink composition of claim 4, wherein said organic group of said pigment is selected from the group consisting of:  $C_6H_4SO_3^-M^+$ ,  $C_6H_4CO_2^-M^+$ ,  $C_6H_4N(CH_3)_3^+X^-$ ,  $C_6H_4COCH_2N(CH_3)_3^+X^-$ , and  $C_6H_4(NC_5H_5)^+X^-$ , wherein  $M^+$  is  $Na^+$ ,  $K^+$ , or  $Li^+$  and  $X^-$  is a monovalent anion.  
15
6. The ink composition of claim 1, wherein said ionically charged pigment is present in an amount between about 1% and about 20%, by weight, of said ink composition.
7. The ink composition of claim 6, wherein said ionically charged pigment is present in an amount between about 2% and about 10%, by weight, of said ink composition.  
20

- 25 -

8. The ink composition of claim 1, wherein said humectant is present in an amount equal or greater than 15%, by weight, of said ink composition.

9. The ink composition of claim 8, wherein said humectant is present in an amount equal or greater than 20%, by weight, of said ink composition.

5 10. The ink composition of claim 9, wherein said humectant is present in an amount ranging between 20%, by weight, and 35% by weight, of said ink composition.

11. The ink composition of claim 1, wherein said humectant is selected from the group consisting of: glycols, glycerol, polyols and derivatives and mixtures thereof.

12. The ink composition of claim 11, wherein said glycol is a ethylene glycol, butylene glycol, propylene glycol, diethylene glycol, dipropylene glycol, pentamethylene glycol, polyethylene glycol, polypropylene glycol, trimethylene glycol and mixtures thereof.

13. The ink composition of claim 11, wherein said polyol is a butanetriol, trishydroxymethylmethane, mesoerythritol, xylitol and mixtures thereof.

15 14. The ink composition of claim 1, wherein said composition further comprises a polymer selected from the group consisting of: polyvinyl alcohol, polyester, polyestermelamine, styrene-acrylic acid copolymers, styrene-maleic acid copolymers, styrene-maleic acid-alkyl acrylate copolymers, styrene-metacrylic acid copolymers, styrene-metacrylic acid-alkyl acrylate copolymers, styrene-maleic half ester copolymers, vinyl naphthalene-acrylic acid copolymers, vinyl naphthalene-maleic acid copolymers and salts and mixtures thereof.

20 15. The ink composition of claim 1, wherein said composition further comprises a polymer selected from the group consisting of: polyvinylimidazole, derivatives of

polyvinylimidazole, copolymers of vinylimidazole, copolymers of vinylimidazole derivatives, polyvinylpyridine, derivatives of polyvinylpyridine, copolymers of vinylpyridine, copolymers of vinylpyridine derivatives, polyethyleneimine, derivatives of polyethyleneime, and mixtures thereof.

- 5        16. The ink composition of claim 1, wherein said ionically charged pigment is a carbon black having  $C_6H_4N(CH_3)_3^+Cl^-$  surface groups and said humectant is glycerol.
- 10      17. The ink composition of claim 1, wherein said ionically charged pigment is a carbon black having  $C_6H_4SO_3^-Na^+$  surface groups and said humectant is glycerol.
- 10      18. The ink composition of claim 1, wherein said ionically charged pigment is a carbon black having  $C_6H_4SO_3^-Na^+$  surface groups and said humectant is ethylene glycol.
- 15      19. The ink composition of claim 1, wherein said composition further comprises an additive selected from the group consisting of: binders, biocides, buffers, drying accelerators, humectants, penetrants, and surfactants.
- 20      20. The ink composition of claim 1, wherein said ink composition is an ink jet ink.
- 20      21. A method for generating printed images which comprises the steps of: incorporating into a printing apparatus the ink composition comprising an ionically charged pigment and a hydroxylated hydrocarbon humectant having at least two hydroxyl groups, wherein said humectant is present in an amount sufficient to improve the latency of said ink composition in a printing apparatus; and generating an image onto a substrate.

- 27 -

22. The method of claim 21, wherein said ink composition is an aqueous ink composition.
23. The method of claim 21, wherein said ionically charged pigment is selected from the group consisting of: black, blue, brown, cyan, green, violet, magenta, red, orange, yellow and mixtures thereof.
- 5
24. The method of claim 21, wherein said ionically charged pigment has at least one attached organic group.
25. The method of claim 24, wherein said organic group of said pigment is selected from the group consisting of:  $C_6H_4SO_3^-M^+$ ,  $C_6H_4CO_2^-M^+$ ,  $C_6H_4N(CH_3)_3^+X^-$ ,  $C_6H_4COCH_2N(CH_3)_3^+X^-$ , and  $C_6H_4(NC_5H_5)^+X^-$ , wherein  $M^+$  is  $Na^+$ ,  $K^+$ , or  $Li^+$  and  $X^-$  is a monovalent anion.
- 10
26. The method of claim 21, wherein said ionically charged pigment is present in an amount between about 1% and about 20%, by weight, of said ink composition.
27. The method of claim 26, wherein said ionically charged pigment is present in an amount between about 2% and about 10%, by weight, of said ink composition.
- 15
28. The method of claim 21, wherein said humectant is present in an amount equal or greater than 15%, by weight, of said ink composition.
29. The method of claim 28, wherein said humectant is present in an amount equal or greater than 20%, by weight, of said ink composition.
- 20
30. The method of claim 29, wherein said humectant is present in an amount ranging between 20%, by weight, and 35% by weight, of said ink composition.

31. The method of claim 21, wherein said humectant is selected from the group consisting of: glycols, glycerol, polyols and derivatives and mixtures thereof.

32. The method of claim 31, wherein said glycol is a ethylene glycol, butylene glycol, propylene glycol, diethylene glycol, dipropylene glycol, pentamethylene glycol, polyethylene glycol, polypropylene glycol, trimethylene glycol and mixtures thereof.  
5

33. The method of claim 31, wherein said polyol is a butanetriol, trishydroxymethylethane, mesoerythritol, xylitol and mixtures thereof.

34. The method of claim 21, wherein said composition further comprises a polymer selected from the group consisting of: polyvinyl alcohol, polyester, 10 polyestermelamine, styrene-acrylic acid copolymers, styrene-maleic acid copolymers, styrene-maleic acid-alkyl acrylate copolymers, styrene-metacrylic acid copolymers, styrene-metacrylic acid-alkyl acrylate copolymers, styrene-maleic half ester copolymers, vinyl napthalene-acrylic acid copolymers, vinyl napthalene-maleic acid copolymers and salts and mixtures thereof.  
15

35. The method of claim 21, wherein said composition further comprises a polymer selected from the group consisting of: polyvinylimidazole, derivatives of polyvinylimidazole, copolymers of vinylimidazole, copolymers of vinylimidazole derivatives, polyvinylpyridine, derivatives of polyvinylpyridine, copolymers of vinylpyridine, copolymers of vinylpyridine derivatives, polyethyleneimine, derivatives of polyethyleneime, and mixtures thereof.  
20

36. The method of claim 21, wherein said ionically charged pigment is a carbon black having  $C_6H_4N(CH_3)_3^+Cl^-$  surface groups and said humectant is glycerol.

- 29 -

37. The method of claim 21, wherein said ionically charged pigment is a carbon black having  $C_6H_4SO_3^-Na^+Cl^-$  surface groups and said humectant is glycerol.

38. The method of claim 21, wherein said ionically charged pigment is a carbon black having  $C_6H_4SO_3^-Na^+$  surface groups and said humectant is ethylene glycol.

5 39. The method of claim 21, wherein said composition further comprises an additive selected from the group consisting of: binders, biocides, buffers, drying accelerators, humectants, penetrants, and surfactants.

40. The method of claim 21, wherein said ink composition is an ink jet ink.

## INTERNATIONAL SEARCH REPORT

International Application No  
PCT/US 97/08049

A. CLASSIFICATION OF SUBJECT MATTER IPC 6 C09D11/00		
According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED		
Minimum documentation searched (classification system followed by classification symbols) IPC 6 C09D		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched		
Electronic data base consulted during the international search (name of data base and, where practical, search terms used)		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	<p>EP 0 475 075 A (XEROX CORPORATION) 18 March 1992</p> <p>see abstract see page 5, line 16-25 see page 6, line 7-19 see page 7, line 49-51 see page 8, line 42-45</p> <p>-----</p>	1-4, 6-12,14, 19-24, 26-32, 34,39,40
<input type="checkbox"/> Further documents are listed in the continuation of box C.		<input checked="" type="checkbox"/> Patent family members are listed in annex.
* Special categories of cited documents :		
<p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p>		
<p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&amp;" document member of the same patent family</p>		
Date of the actual completion of the international search	Date of mailing of the international search report	
4 September 1997	15.09.97	
Name and mailing address of the ISA	Authorized officer	
European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax (+31-70) 340-3016	Girard, Y	

**INTERNATIONAL SEARCH REPORT**

Information on patent family members

International Application No

PCT/US 97/08049

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 475075 A	18-03-92	US 5281261 A CA 2046571 A DE 69114962 D DE 69114962 T JP 4234467 A	25-01-94 01-03-92 11-01-96 18-04-96 24-08-92